An interactive web native textbook for Material and Energy Balances: Student Feedback and Performance

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Acknowledgments: Charles Vestal, Marc Donnelly, Yusef Ben-Masuad, Brian Yoon, Jake Newsom, David C. Smith, zyBooks team, countless TAs



Disclaimer: I may financially benefit from sales of the book discussed in this talk

What fraction of your students read the textbook?

Does reading a textbook translate into higher course grades?

Interactive technology beneficial



Internet >> Textbook

Interactive web-native > static web

Students want interactive books

A short list of references:

T. Stelzer, et al., Am. J. Phys. 2009.

C. S. Lee, et al., J. Eng. Ed. 2013.

Edgcomb & Vahid, ASEE Conf. Proc. 2014 and 2015 and 2016.

Falconer & Nicodemus, CEE 2014 and ASEE Conf. Proc. 2015.

Branch & Butterfield, ASEE Conf. Proc. 2015 and 2016.

Seeling. FIE 2015.

Vestal, C.R. CEE Teaching Tip 2016.

A common theme for talks in this session

Setting goals



MEB course goal:

Develop problem solving skills and conceptual understanding

'Textbook' authoring goal:

Bring active learning to a textbook

Features of a zyBook



Definitions

Learning questions

T/F, Multiple choice, Matching, Short answer

Animations

Links to additional, external resources

Challenge activities

Less text, more actionTM

Looking into the MEB zyBook



Material and Energy Balances

Quantities, Units, and Calculations
Material Balances
Reacting Systems
Solids, Liquids, and Gases
Multiphase Systems
Energy Balances
Transient Systems

zyBooks.com

Less text, more actionTM



Entering a reactor are water at 44 mol/hr and carbon at 36 mol/hr to complete the following reaction: $C + 2 H_2O \rightarrow CO_2 + 2 H_2$. If 28 mol/hr of hydrogen are formed in the reactor, the fraction conversion of carbon should be:	0.78
	0.39



Entering a reactor are water at 44 mol/hr and carbon at 36 mol/hr to complete the following reaction:

 $C + 2 H_2O \rightarrow CO_2 + 2 H_2$.

If 28 mol/hr of hydrogen are formed in the reactor, the fraction conversion of carbon should be:

O.78 comes from taking the ratio of 28 mol/hr of hydrogen in the reactor product to the 36 mol/hr of carbon fed to the reactor. While the 36 mol/hr of carbon fed is correct, the numerator of the fractional conversion should be a flow rate of carbon.

0.78		
0.64		
0.39		



Entering a reactor are water at 44 mol/hr and carbon at 36 mol/hr to complete the following reaction:

 $C + 2 H_2O \rightarrow CO_2 + 2 H_2$.

If 28 mol/hr of hydrogen are formed in the reactor, the fraction conversion of carbon should be:

★ 0.64 is the correct conversion of water, which would come from a ratio of 28 mol/hr of water reacted to 44 mol/hr of water fed. However, the question asks for the conversion of carbon, not water. 0.78 0.64 0.39



Entering a reactor are water at 44 mol/hr and carbon at 36 mol/hr to complete the following reaction:

 $C + 2 H_2O \rightarrow CO_2 + 2 H_2$.

If 28 mol/hr of hydrogen are formed in the reactor, the fraction conversion of carbon should be:

✓ Using a stoichiometric ratio, 1 mol of C reacts to form 2 mol of H₂. Therefore, 14 mol/hr of carbon reacted. Then the fraction conversion would be 14 mol/hr / 36 mol/hr = 0.39.

0.78

0.64

0.39

Over 75 animations



Turn static derivations into live exercises Simplifying energy balance

Build up static figures

Constructing a phase diagram

Perform physical processes in equipment

Operating a distillation column

95% of students report watching multiple times

How much textbook reading occurs?



25-30% of students self reporting

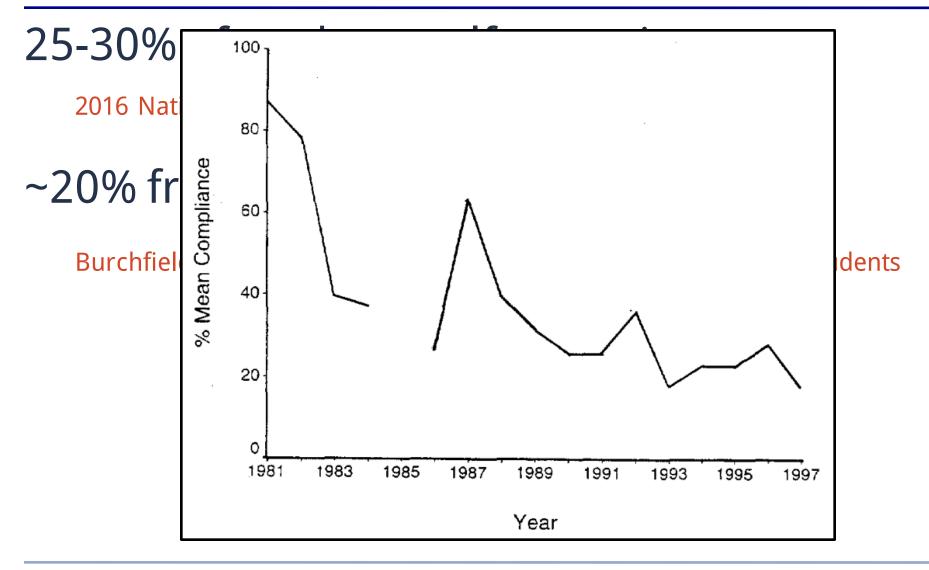
2016 National Survey of Student Engagement. > 280,000 students

~20% from self report/quiz grade study

Burchfield and Sappington, *Teaching of Psychology*, 2000. > 900 students

How much textbook reading occurs?





Very few studies/data exist

Providing an incentive



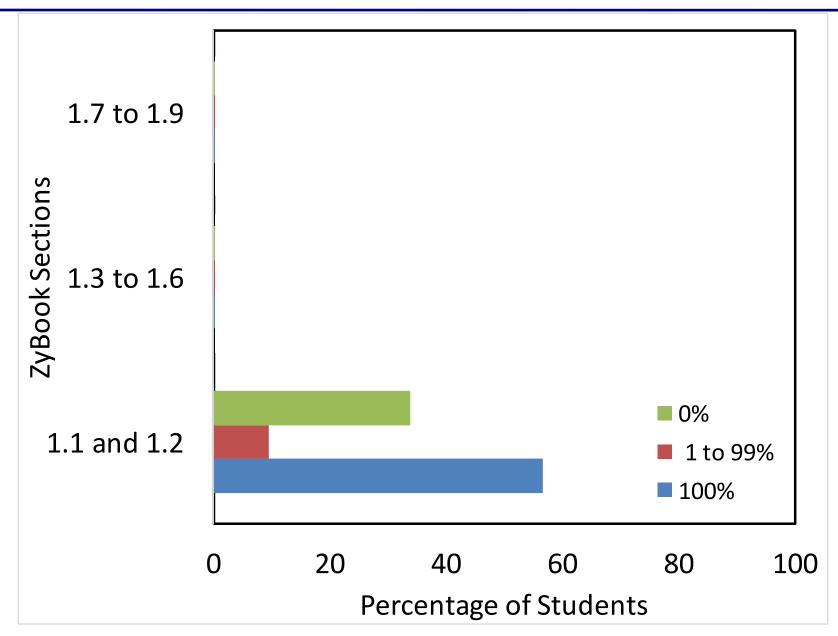
Offering 2 to 10% of final course grade leads to high reading rates

Edgcomb and Vahid. FIE. 2015. > 1100 students

6% of final course grade offered for MEB

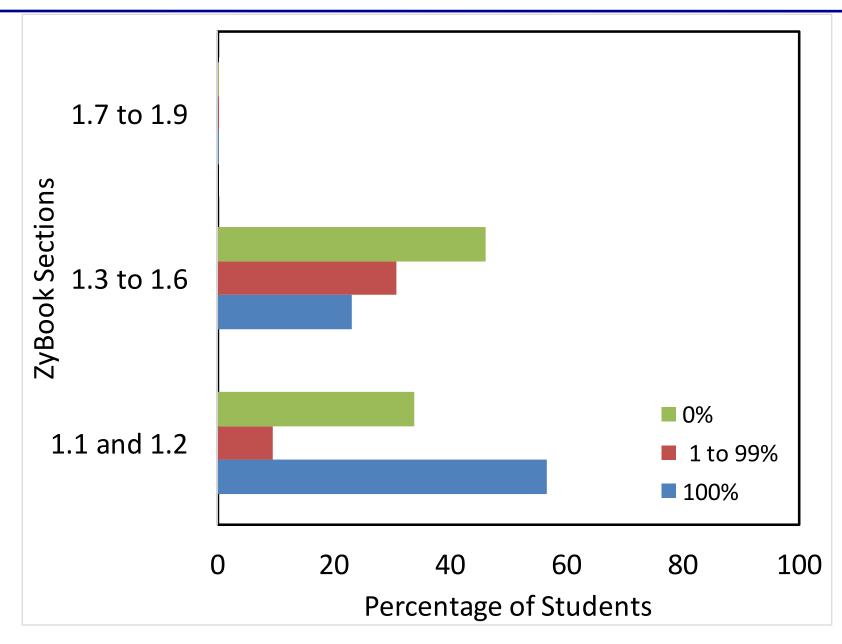
Learning to read/participate





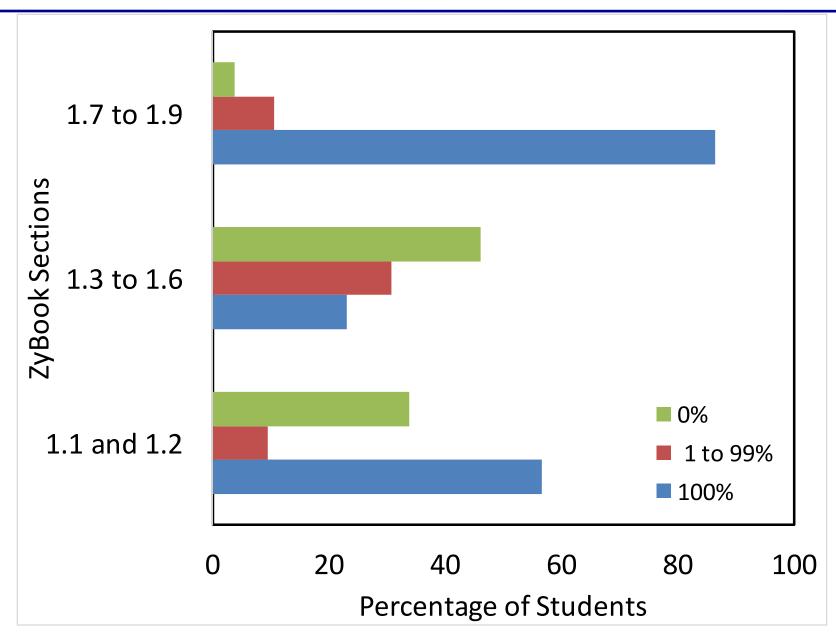
Learning to read/participate





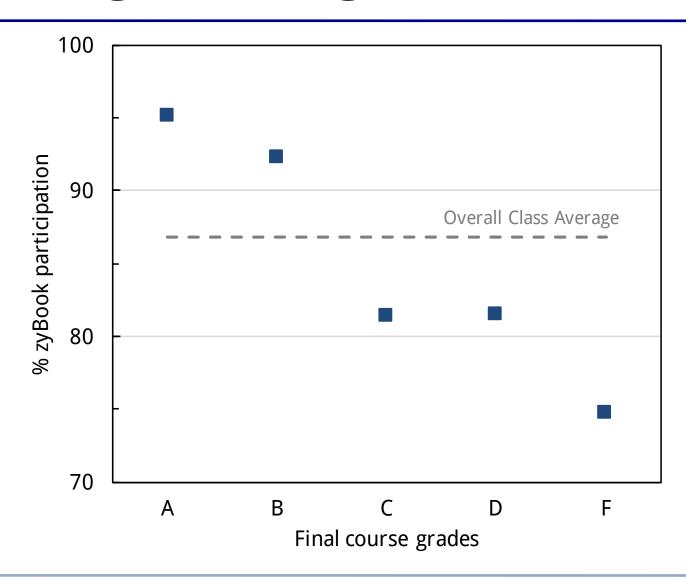
Learning to read/participate





87% average reading rate

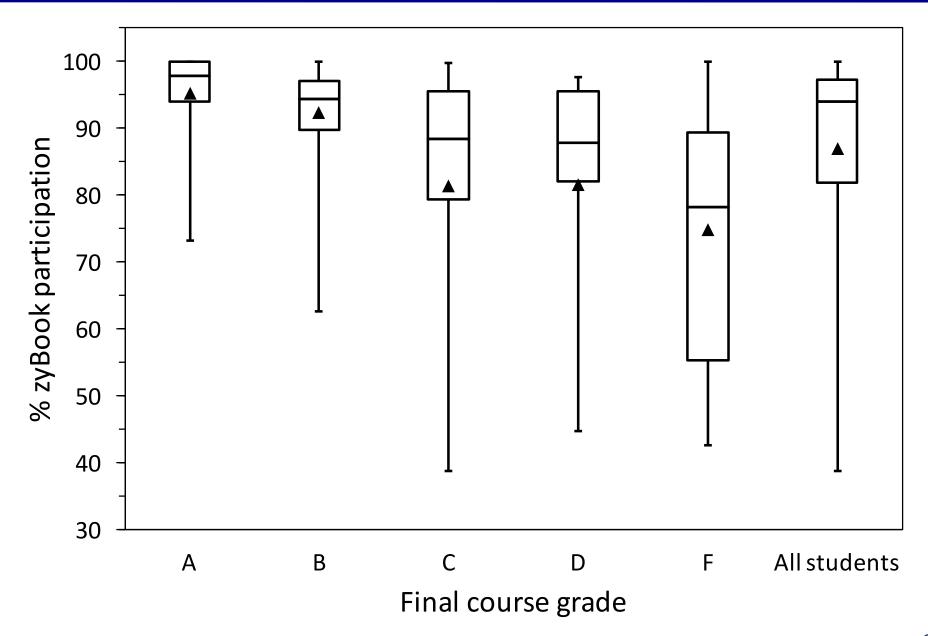




Read by due date + 67 sections + 100 students

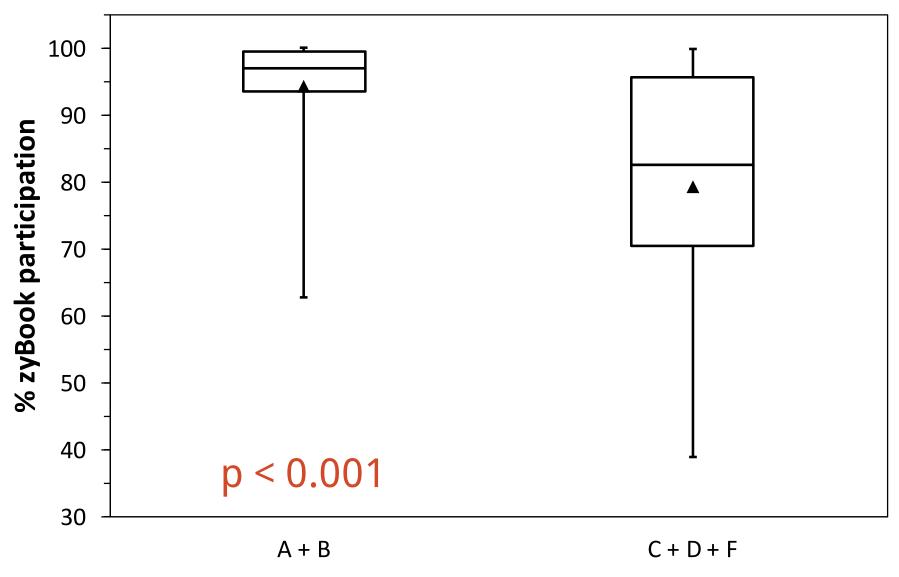
Reading correlates with final grades





A,B students read a lot more

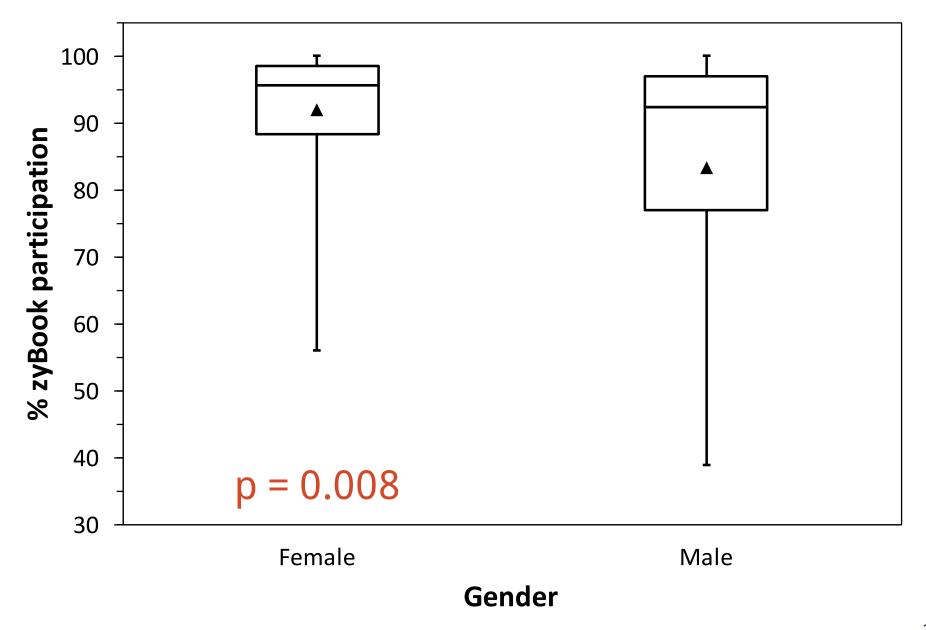




Final course grades

Females read more





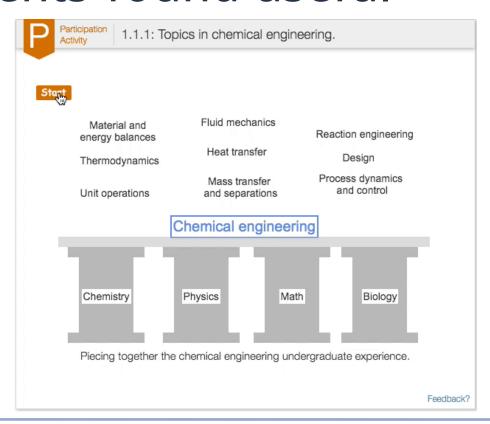
Zybook has high reading rate



of zyBook read of students found useful

Reading correlated with final grades

Zybook preferred to static book



Better things will keep coming - zyBooks.com